

D. Applications of Tunneling

(a) α -decay in NucleiData of 5 α -particle emitting nuclei

Nucleus	k.e. (MeV)	half-life $t_{1/2}$
^{216}Ra	9.5	0.18 μs
^{194}Po	7.0	0.7 s
^{240}Cm	6.4	27 days
^{226}Ra	4.9	1600 years
^{232}Th	4.1	14 billion years

α -particle
k.e. don't vary much!

$t_{1/2}$ vary by a lot!

↑
Many orders
of magnitude

↓

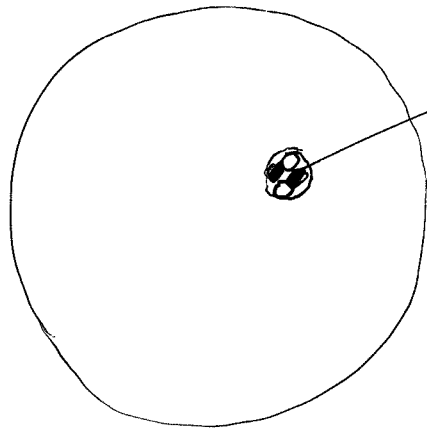
Note correlation:

longer $t_{1/2} \leftrightarrow$ smaller α -particle k.e.

shorter $t_{1/2} \leftrightarrow$ higher α -particle k.e.

Back to tunneling and α -decay

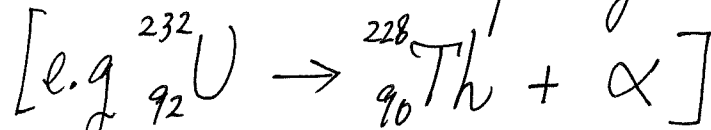
- Gamov (1928) : α -decay comes from tunneling



${}^4_2\text{He}$ "grouped" occasionally

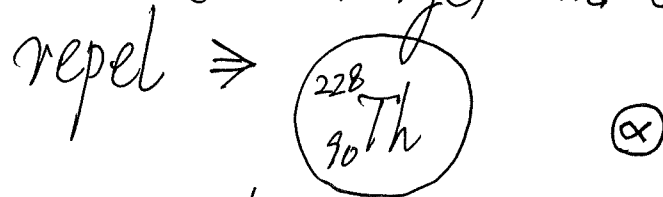
- travels toward surface

\Rightarrow an attempt of tunneling

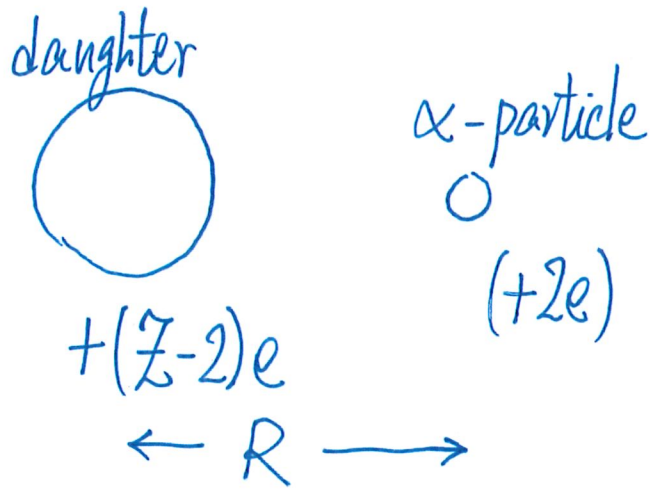


- But other nucleons (${}^{228}_{90}\text{Th}$) exert attraction (nuclear force) and don't want α to leave \Rightarrow a barrier

- Once outside : α (+ve charge) and daughter nucleus (+ve charge)

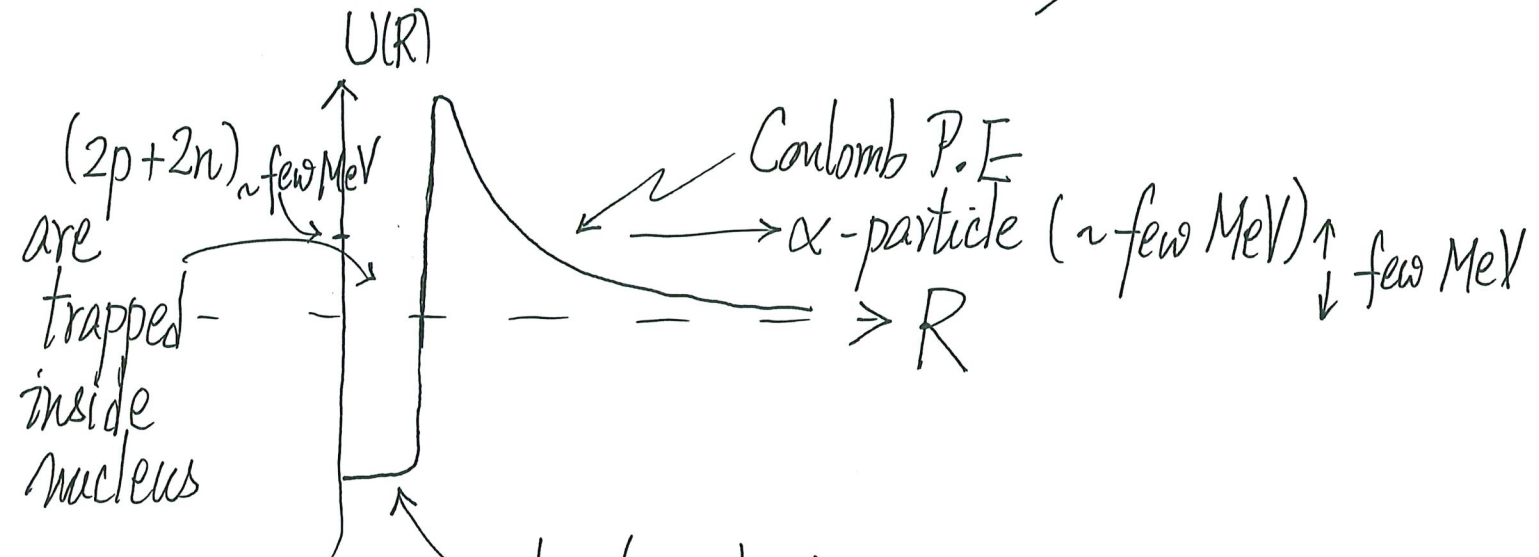


don't want α to come back



Coulomb repulsion

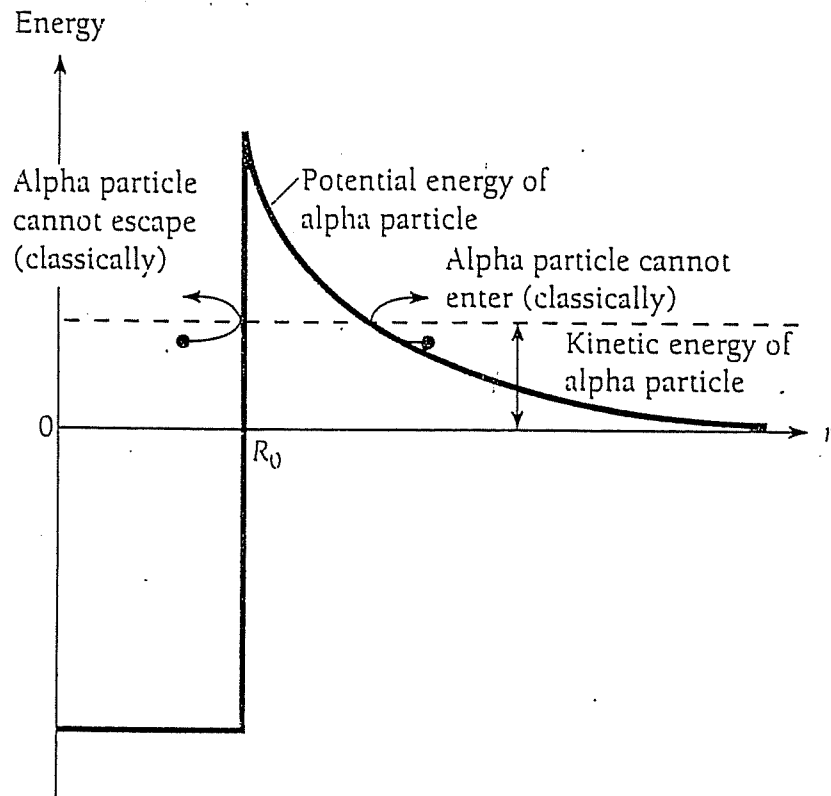
$$[P.E. = + \frac{(Z-2) \cdot 2e^2}{4\pi\epsilon_0 R}]$$



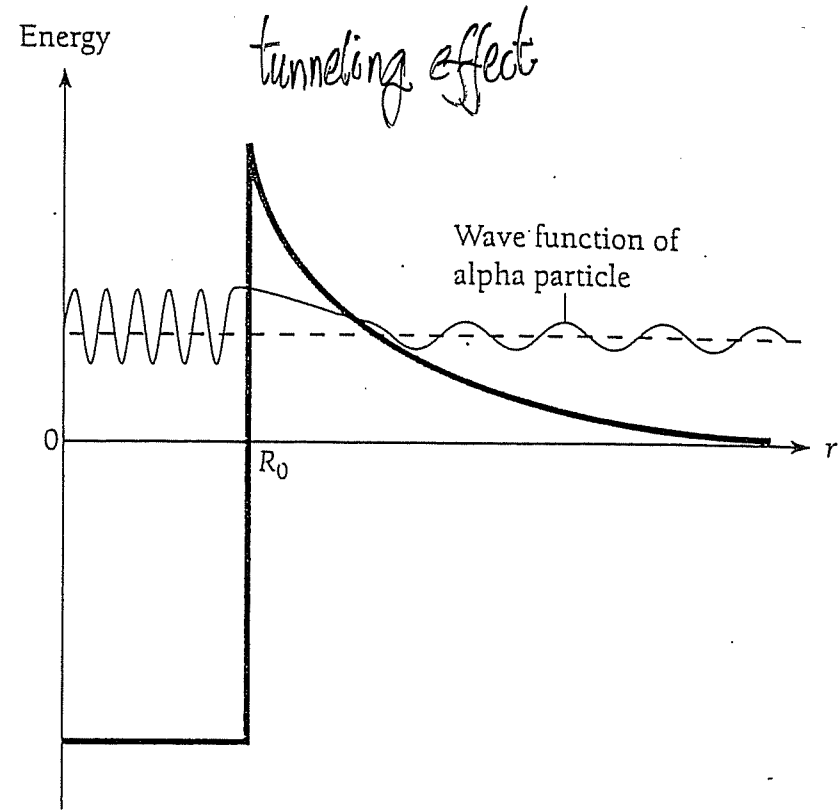
due to short-range ($\sim 10^{-15}$ m) nuclear force

Q: Does α -particle climb out of the barrier?
 [No! α -particle k.e. is NOT that high!]

Q: How does α -decay occur? [Tunneling]



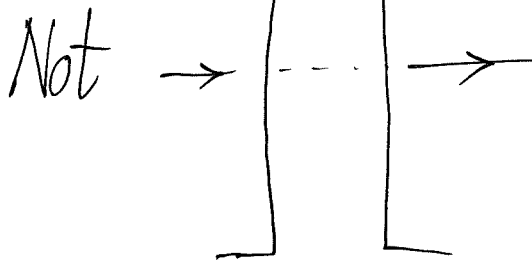
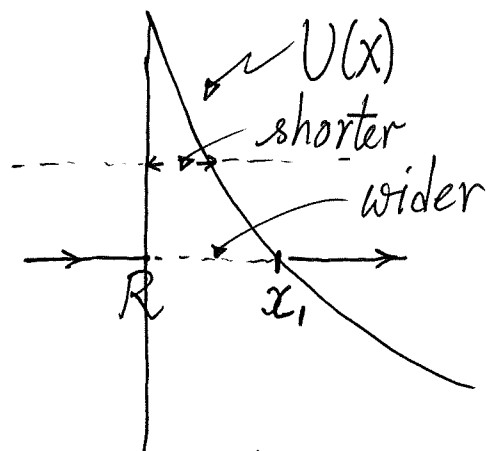
(a)



(b)

(a) In classical physics, an alpha particle whose kinetic energy is less than the height of the potential barrier around a nucleus cannot enter or leave the nucleus, whose radius is R_0 . (b) In quantum physics, such an alpha particle can tunnel through the potential barrier with a probability that decreases with the height and thickness of the barrier.

This is why α -particles (a few MeV) can be detected even the barrier is higher.



$$\sim e^{-2 \int_R^{x_1} \sqrt{\frac{2m}{\hbar^2} (U(x) - E)} dx}$$

$$\sim e^{-2L \sqrt{\frac{2m}{\hbar^2} (U - E)}}$$

[slight increase in E changes the value of T by orders of magnitude!]

Ex. $\frac{Z}{L} = 90$, $E \sim 6 \text{ MeV}$, $R \sim 8 \text{ fm}$

$T \sim 10^{-29}$ (tiny!)

prob. of getting through per attempt (interpretation)

- So $T \sim e^{-2L\sqrt{\frac{2m}{\hbar^2}(U-E)}}$ is tiny. Why could we see α -decays?
 \hookrightarrow prob. of getting through per attempt

- Although T is tiny, α -particle keeps on trying!

Estimate: Diameter of nucleus = $D \sim 15$ fm (big nucleus)
 α -particle's speed inside nucleus $\sim c/10$ (roughly)

$$f = \# \text{ attempts per second} \sim \frac{c/10}{15 \text{ fm}} \sim \underbrace{10^{21}}_{\text{try } 10^{21} \text{ times per second}} \text{ s}^{-1}$$

$$\therefore \lambda = \frac{\text{Prob. of a nucleus to decay per second}}{\tau} = \underbrace{T}_{\substack{\text{tiny} \\ \text{(tunneling)}}} \cdot \underbrace{f}_{\substack{\text{big number} \\ \text{(s}^{-1}\text{)}}} \quad (\tau \cdot \ln 2 = t_{1/2})$$